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Key indicators

Single-crystal X-ray study T = 292 KMean σ (C–C) = 0.002 Å R factor = 0.051 wR factor = 0.157 Data-to-parameter ratio = 14.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N,*N*'-Diphenylethylenediamine, an N— $H \cdots \pi$ - and C— $H \cdots \pi$ -bonded network

N,*N*'-Diphenylethylenediamine, $C_{14}H_{16}N_2$, forms monoclinic crystals, with a centre of symmetry at the mid-point of the central C–C bond. In the crystal structure, the intermolecular interactions are primarily of N–H··· π and C–H··· π character, without any contribution from N–H···N hydrogen bonding. The N–H··· π and C–H··· π interactions, when considered separately, give rise to two sets of layers that, on superposition, generate a network structure.

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Comment

There are three crystal structures reported in the literature containing N, N'-diphenvlethylenediamine ligands, as revealed by a search in the Version 5.26 of the Cambridge Structural Database (Allen, 2002). In all of these cases, the N atoms display tetrahedral geometries and are chirogenic. Crystal structures of compounds displaying stereochemically labile chirogenic atoms are interesting when considering possible candidates for absolute asymmetric synthesis (Vestergren et al., 2003; Lennartson et al., 2005). This, however, requires that the compound crystallizes in one of the Sohncke space groups. In two of the structures, [Ni(acac)NO₃(dpheda)₂] (Voutsas, 1995), where acac is acetylacetonate and dpheda is N,N'-diphenylethylenediamine, and [Ni(hfac)₂(dpheda)₂] (Voutsas et al., 1995), where hfac is hexafluoroacetylacetonate, the two N atoms in each ligand molecule have the same configuration, whereas the third structure, [Ni(dpheda)₂tfac]ClO₄ (Voutsas et al., 1994), where tfac is 1,1,1-trifluoroacetylacetonate, is a meso form. We now report the crystal structure of the free ligand N, N'-diphenylethylenediamine, (I).



Since compound (I) (Fig. 1) crystallizes in the centrosymmetric space group $P2_1/c$ with the $C7-C7^i$ ethane bond situated on an inversion centre [symmetry code: (i) 1 - x, -y,-z], it follows that (I) displays a perfectly staggered conformation in the solid state, with an $N1-C7-C7^i-N1^i$ torsion angle of 180° . The coordination geometry around the N atoms is perhaps best described as trigonal planar, with C7-N1-

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Figure 1

ORTEP-3 (Farrugia, 1997) plot of (I), with the atom-numbering scheme. Displacement ellipsoids drawn at the 50% probability level. All H atoms have been omitted. [Symmetry code: (i) 1 - x, -y, -z.]



Figure 2 $N-H\cdots\pi$ interactions (dashed lines) viewed along the *a* axis.

H1, C1-N1-H1 and C7-N1-C1 angles of 119 (2), 117 (2) and 125.2 (2)°, respectively. The molecule does not, in other words, display any element of chirality.

The crystal structure displays two distinct types of intermolecular interactions, $N-H\cdots\pi$ and $C-H\cdots\pi$ interactions. Interestingly, there are no $N-H \cdots N$ hydrogen bonds, which gives an indication on the importance of $C-H\cdots\pi$ and N- $H \cdots \pi$ interactions in the assembly of discrete molecules into crystal structures (Nishio, 2004; Cantrill et al., 2000; Braga et al., 1998; Viswamitra et al., 1993). The N-H $\cdots\pi$ interactions depend on the interactions between the NH group and the π electrons in the C3-C4 bond of an adjacent molecule. The H1···C3ⁱⁱ and H1···C4ⁱⁱ [symmetry code: (ii) 1 - x, $y - \frac{1}{2}$, $\frac{1}{2} - z$] distances are 2.73 (2) and 2.83 (2) Å, respectively, and the N1-H1···C3ⁱⁱ and N1-H1···C4ⁱⁱ angles are 170 (2) and 143 (2)°, respectively. The N-H··· π interactions extend the structure only in the bc plane; there are no N-H··· π interactions in the *a*-axis direction. If only the N-H··· π interactions are considered infinite layers are formed, where each molecule interacts with four neighbours, as shown in Fig. 2.

The C-H··· π interactions (Fig. 3) involve the H3 atom and the C1-C6 bond in an adjacent molecule, with a H3···C6ⁱⁱⁱ distance of 2.83 Å, and a H3···C1ⁱⁱⁱ distance of 2.99 Å [symmetry code: (iii) $2 - x, \frac{1}{2} + y, \frac{1}{2} - z$]. The C3-H3···C6ⁱⁱⁱ and C3-H3···C1 angles are 167 and 144°, respectively. In total, every molecule is involved in four interactions of this kind. Like the N-H··· π interactions, the C-H··· π -interactions propagate in two dimensions, giving rise to layers. These layers are different from those formed by the N-H··· π interactions, and are extended in the (204) set of planes.



 $C-H\cdots\pi$ interactions (dashed lines) viewed along the *a* axis.





Intersection between N-H··· π bonded layers (horizontal) and C-H··· π -bonded layers viewed along the *b* axis.

The two different sets of interactions thus give rise to two different sets of layers intersecting at an angle of 37° (Fig. 4). Since each molecule is a part of both kinds of layers, a three-dimensional network results, with no overall layered structure.

Experimental

Commercial (E. Merck) N,N'-diphenylethylenediamine (0.50 g) was heated with hexane (3 ml) to gentle reflux, and toluene was added dropwise until dissolution. The clear solution was slowly allowed to reach ambient temperature, whereupon large colourless plates formed. Samples for crystallographic analysis were cleaved from larger crystals.

Crystal	data
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$C_{14}H_{16}N_2$	$D_x = 1.206 \text{ Mg m}^{-3}$
$M_r = 212.29$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 3912
a = 5.883 (2) Å	reflections
b = 8.022 (3) Å	$\theta = 3.0-26.0^{\circ}$
c = 12.387(5) Å	$\mu = 0.07 \text{ mm}^{-1}$
$\beta = 90.185 \ (15)^{\circ}$	T = 292 (2) K
$V = 584.6 (4) \text{ Å}^3$	Plate, white
Z = 2	$0.40 \times 0.25 \times 0.25 \text{ mm}$
Data collection	
Rigaku R-AXIS IIc image-plate	848 reflections with $I > 2\sigma(I)$
system diffractometer	$R_{\rm int} = 0.055$
o scans	$\theta^{} = 26.0^{\circ}$

Absorption correction: none 3912 measured reflections

1116 independent reflections

 $15 \rightarrow 15$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0963P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.051$	+ 0.0225P]
$wR(F^2) = 0.157$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.00	$(\Delta/\sigma)_{\rm max} = 0.001$
1116 reflections	$\Delta \rho_{\rm max} = 0.15 \ {\rm e} \ {\rm \AA}^{-3}$
77 parameters	$\Delta \rho_{\rm min} = -0.11 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected geometric parameters (Å, °).

C1-N1	1.373 (2)	C5-C4	1.378 (3)
C1-C2	1.390 (2)	C4-C3	1.367 (3)
C1-C6	1.392 (2)	N1-C7	1.441 (2)
C6-C5	1.359 (2)	$C7-C7^{i}$	1.510 (3)
C2-C3	1.375 (2)		
N1-C1-C2	122.80 (14)	C1-N1-C7	125.19 (16)
N1-C1-C6	119.24 (15)		

Symmetry code: (i) 1 - x, -y, -z.

All H atoms except H1 were included in calculated positions (C– H = 0.96–0.97 Å) and refined using a riding model, with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$. Atom H1 was located in a difference map and allowed to refine without constraints.

Data collection: *CrystalClear* (Rigaku, 2000); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine

structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Altomare, A., Cascarano, G., Giacovazzo, C. & Gualardi, A. (1993). J. Appl. Cryst. 26, 343–350.
- Braga, D., Grepioni, F. & Tedesco, E. (1998). Organometallics, 17, 2669–2672. Cantrill, S. J., Preece, J. A., Stoddart, J. F., Wang, Z. H., White, A. J. P. &
- Williams, D. J. (2000). Tetrahedron, 56, 6675–6681.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Lennartson, A., Vestergren, M. & Håkansson, M. (2005). Chem. Eur. J. 11, 1757–1762.
- Nishio, M. (2004). CrystEngComm, 6, 130-158.
- Rigaku (2000). CrystalClear. Version 1.3. Rigaku Corporation, Tokyo, Japan.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen. Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7–13.
- Vestergren, M., Eriksson, J. & Håkansson, M. (2003). Chem. Eur. J. 9, 4678–4686.
- Viswamitra, M. A., Radhakrishnan, R., Bandekar, J. & Desiraju, G. R. (1993). J. Am. Chem. Soc. 115, 4868–4869.
- Voutsas, G. P. (1995). Z. Kristallogr. 210, 31-33.
- Voutsas, G. P., Keramidas, K. G., Tsiamis, C. & Themeli, M. (1994). Z. Kristallogr. 209, 965–968.
- Voutsas, G., Keramidas, K., Tsiamis, C. & Themeli, M. (1995). Z. Kristallogr. **210**, 107–110.